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Chemistry – Unit 6 Sticky Tape Post-Lab Thomson Model and Sticky Tape Let's see how we can use Thomson's model to explain the behavior of the sticky tape when we made our tape stacks. A few atoms from the top tape and the bottom tape are represented in the diagram below. Add electrons to each atom to show what happens to the electrons when we make a tape stack out of neutral pieces of tape and then pull them apart.

1 *Sticky Tape*

Modeling Instruction – AMTA 2013 1 U6 Sticky Tape v1.0 Chemistry – Unit 6 Sticky Tape Post-Lab Thomson Model and Sticky Tape Let's see how we can use Thomson's model to explain the behavior of the sticky tape when we made our tape stacks. A few atoms from the top tape and the bottom tape are represented in the diagram below.

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Sticky Tape lab summary. This experiment that objects with the same charge repeal each other and objects with opposite charges attract each other. The top tape was positive and the bottom tape was negative. Thomson's model of the Atom. Also called the plum pudding model.

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The ones on the right are next to a top tape (+ charge). Modeling Instruction – AMTA 2013 1 U6 Sticky Tape v1.0 bottom (-) Before adhesion In contact Separated top(+) Add force vectors to the non-metal (paper) atoms and the top tape in the first row to show the attraction between the paper and the tape.

02bstickytapemodelnotes-v2.doc - Lets see how we can use ...

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2 Sticky Tape Post Lab Answers Unit 2 Sticky Tape Post Use efficient and reliable sticky notes to organize tasks, brainstorm ideas, color-code projects, and much more. The adhesive notes stick to soft and hard surfaces, such as file cabinets and school folders, and the paper units are available in a variety of colors to brighten workspaces and ...

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Unit 2 Sticky Tape Post Lab Answers

Chemistry 1 U6 Review 2013 Name Date Pd Chemistry – Unit 6 – Describing Compounds Review Part 1 – Unit 6 Concepts 1. Recall your representation of the atoms in the Sticky Tape activity. Below is a pair of tapes before they have been pulled apart. Explain why they would not exert a force (either attractive or repulsive) on one another. 2. Below are groups of the inner cores of the atoms ...

U6 Review - Name Date Pd Chemistry Unit 6 Describing ...

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1 Sticky Tape - University of Kentucky When 3M HCl is added to solid sodium carbonate, the contents of the test tube immediately starts bubbling and gets warm. Carbon dioxide gas, water vapor and sodium chloride are formed. 5. As solid ammonium nitrate dissolves in water, the resulting solution cools. Chemistry Unit 1 Worksheet 6 Answers

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