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Chapter 16 - The Process of Chemical Reactions

Chapter 16 – The Process of Chemical Reactions 249 Exercise 16.5 – Predicting the Effect of Disruptions on Equilibrium: Nitric acid can be made from the exothermic reaction of nitrogen dioxide gas and water vapor in the presence of a rhodium and platinum catalyst at 700-900 C and 5-8 atm. Predict whether each of the following changes in the equilibrium system will shift the system to more products, to more reactants, or to neither.

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562 Chapter 16 • Reaction Rates You can also choose to state the rate of the reaction as the rate at which CO is consumed, as shown below. average reaction rate = $\frac{[\text{CO}] \text{ at time } t_2 - [\text{CO}] \text{ at time } t_1}{t_2 - t_1} = ?$ Do you predict a positive or a negative value for this reaction rate?

Chapter 16: Reaction Rates

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come together to act out emotions – so intensely focused that they

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Reaction Rates Answer Key

may become violent.. Mob – highly emotional crowd engaging in violence against specific target. Riot – violent crowd fueled by deep-seated emotion but not at a specific target. Panic – large number of people react with strong emotions and self-destructive behavior to a real or perceived threat.

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After completing this chapter, you should be able to. identify, by name and symbol, the principal chemical elements of the human body. describe the basic structure of an atom, ion, molecule and compound. explain, briefly, how ionic, covalent and hydrogen bonds are formed. define the term chemical reaction, and explain the basic differences between synthesis, decomposition, exchange and reversible reactions.

Chapter 2 :: BIOL 235 Study Guide

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