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of Chemical
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Chemical equilibrium

with real examples Le

Chatelier's Principle and

Temperature Changes

(Pt. 10)

Le Chatelier's Principle

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~~2--Calculating~~ ~~Chemical~~

~~Equilibrium~~ **The**
Equilibrium Constant

*Chemical Equilibria
and Reaction Quotients*

Which way will the
Equilibrium Shift? (Le
Chatelier's Principle)

*Chemical Equilibrium
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Equilibrium -1

**Physical processes:
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Chemical Equilibrium

chemical equation for
the reaction at

equilibrium is written
using double arrows to

indicate the overall
reversibility of the

reaction. $2\text{HgO}(\text{s}) \rightleftharpoons$

$2\text{Hg}(\text{l}) + \text{O}_2(\text{g})$

Equilibrium, a Dynamic

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State Many chemical reactions are reversible under ordinary conditions of temperature and concentration. They will reach a state of equilibrium

CHAPTER 18

Chemical Equilibrium

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Chapter 18 Chemical

Equilibrium Figure 18-3

If the only way in or out of San Francisco and Sausalito is across the Golden Gate Bridge, which joins the two cities, then the number of vehicles in the two cities will remain constant if the number of vehicles per hour crossing the bridge in one

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CHAPTER 18:

Chemical Equilibrium 1.

a. Write the chemical equilibrium expression for the following

equations. Include the

value of K . (a) $(g) \rightleftharpoons (g)$

$K = 0.1$ (b) $(g) + 0.2 (g)$

$(g) \rightleftharpoons (g)$ $K = 1.8 \times 10^{-2}$ Does

the reaction favor

products or reactants?

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(Will there be mostly

products or reactants

when it reaches 2. a.

Compare the rates of

forward and reverse

reactions when

equilibrium has been

reached. b.

North St. Paul-

Maplewood Oakdale /

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