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1 1.1 Basic concepts: atoms The notation: $^{50}_{24}\text{Cr}$ shows that the atomic number, Z, is 24 and the mass number for the isotope is 50. Number of protons = Number of electrons = Z = 24 Number of neutrons = Mass number - Z = 50 - 24 = 26 For each isotope, Z = 24 and so there are 24 electrons and 24 protons.

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